

Novel Materials for Energy Storage and Conversion



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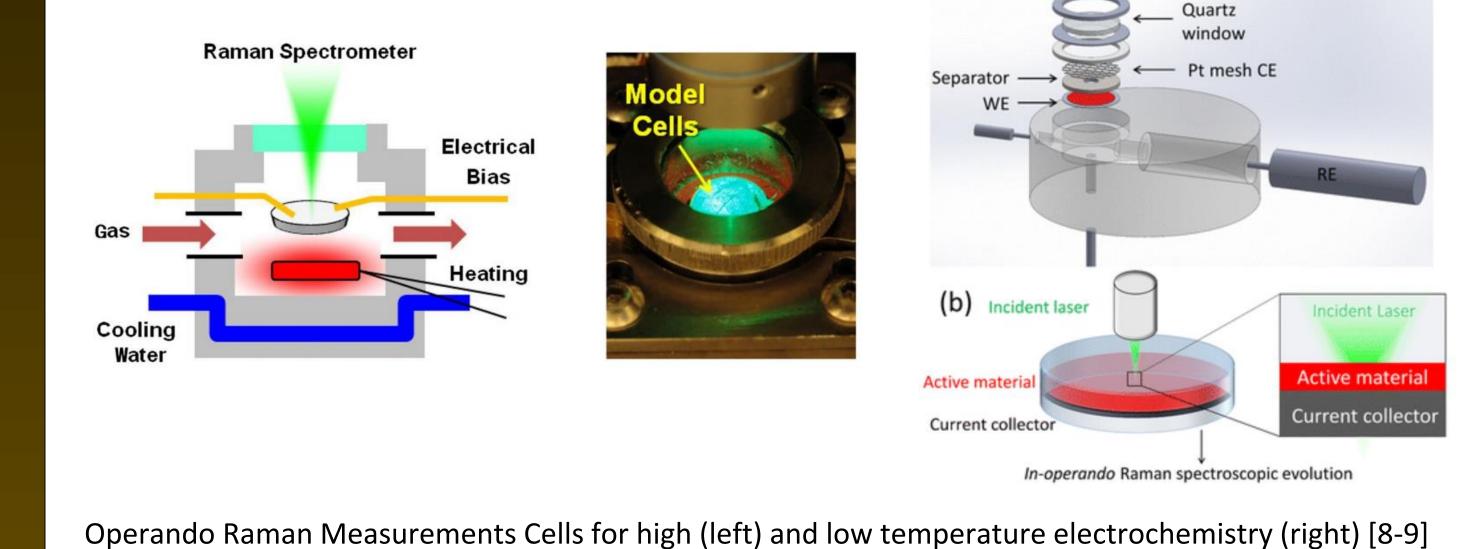


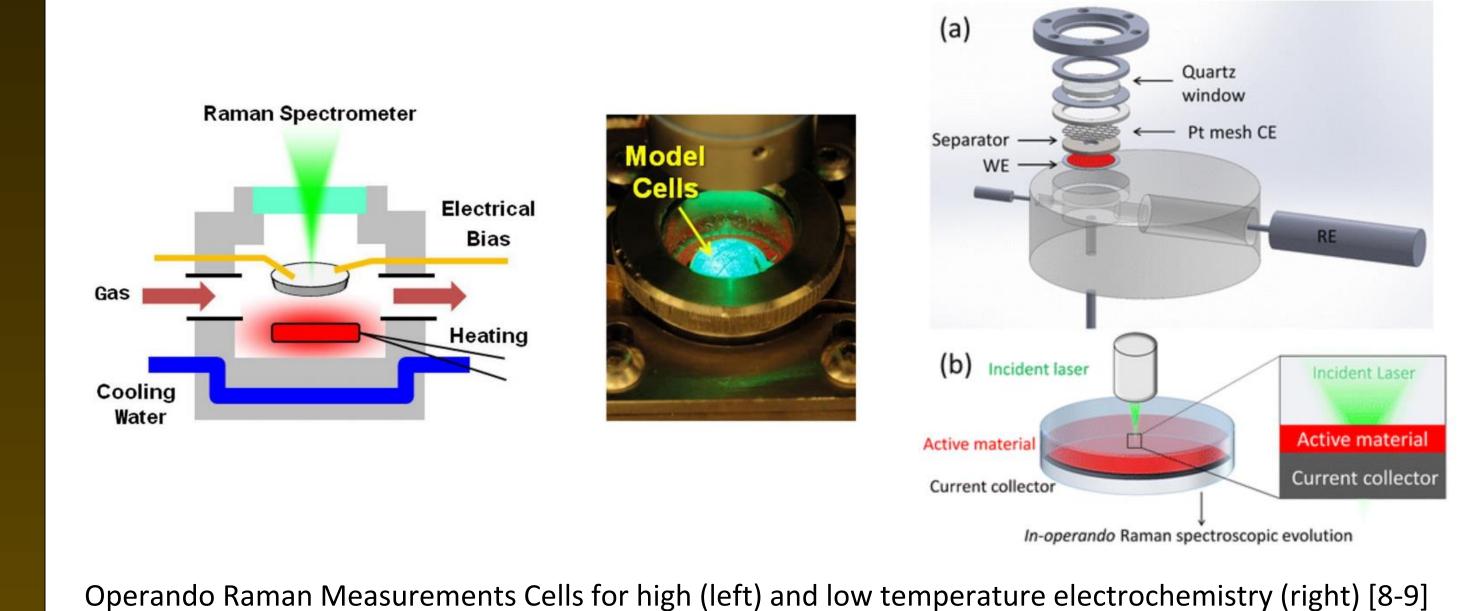
Societal Context

- Electrical Energy Storage & Conversion (EESC) systems are key for the development of technologies that are **efficient and sustainable**
- The performance of EESC systems depends on the development of **new materials**/nanostructures with dramatically enhanced ionic/electronic conductivity and catalytic activities
- Advanced EESC systems can enable the deployment of **higher** efficiency and greener methods for chemical conversion and energy storage for applications like electric vehicles, mobile devices, grid-

Operando and In-Situ Analysis

Use of *Operando* and *in-situ* techniques enables the collection of *more realistic* data and simplifies complex phenomenon using *model systems*





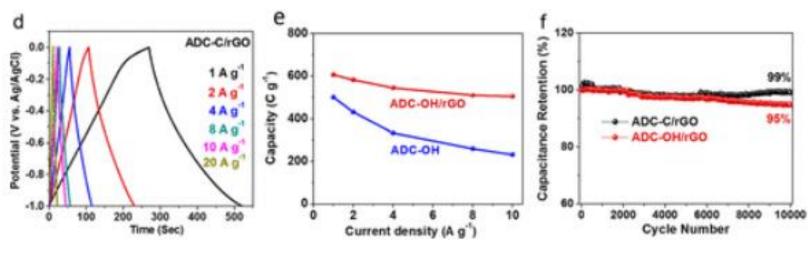
scale energy storage, and chemical production

Electrochemical Materials R&D

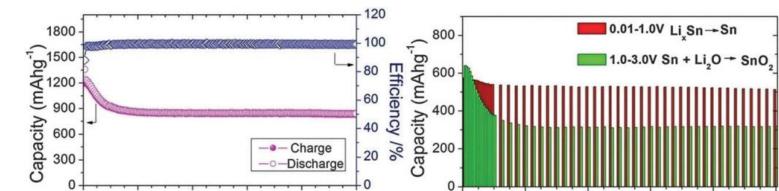
- The discovery of novel materials/nanostructures are at the center of the creation of next-generation energy storage & conversion systems (e.g., batteries, fuel cells, and supercapacitors) for renewable energy
- The rates and efficiency of many chemical and energy transformation processes are determined by materials (catalysts, electrodes, etc...)

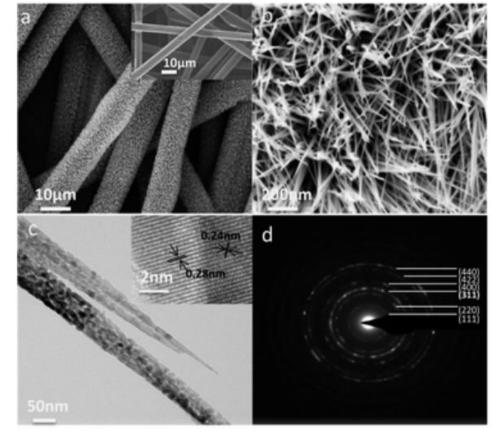
Batteries and Supercapacitors

The improvement of battery and supercapacitor technology requires the development of **new materials and electrode architectures**

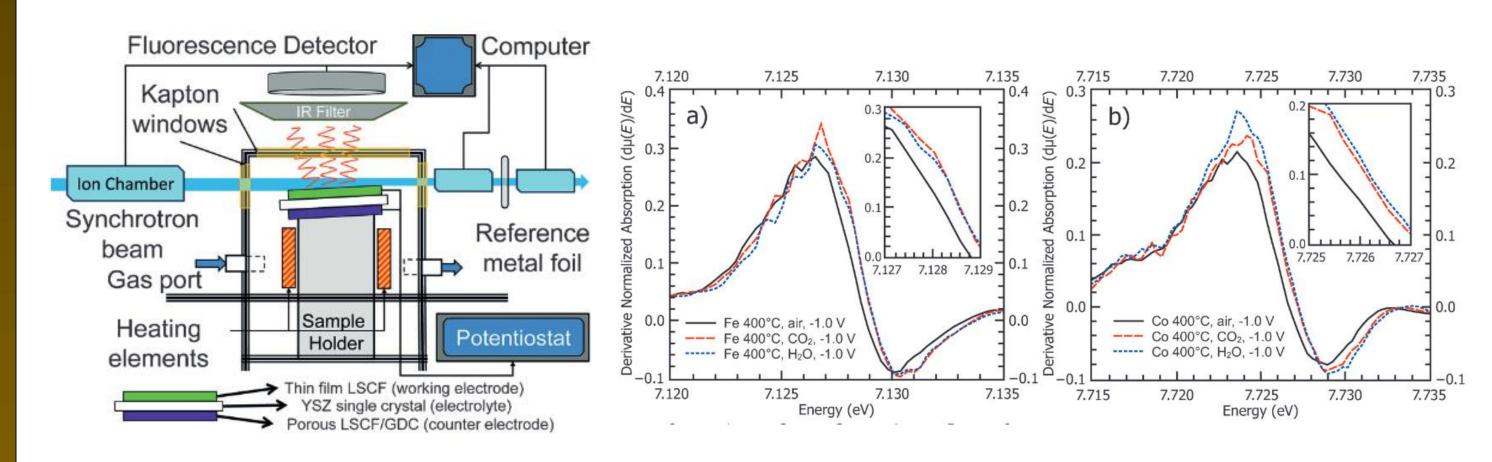


Ni(OH) MOF/graphene composite for supercapacitors [2]

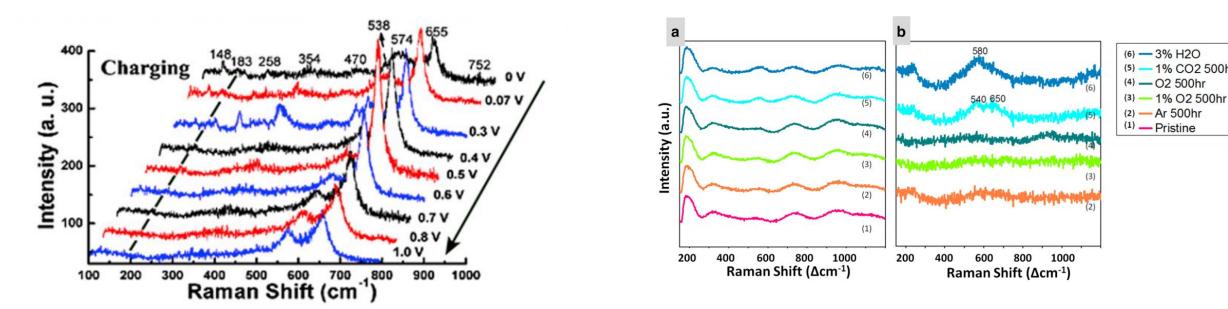




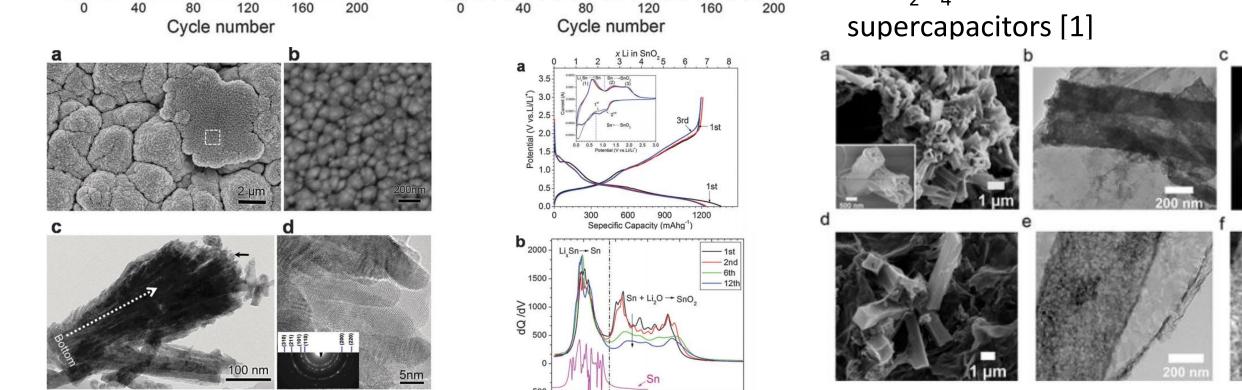
 $NiCo_2O_4$ Nanowires on Carbon Paper for



Operando XAS set-up (left) and corresponding operando XANES of LSCF (right) [10]



Operando low temperature Raman of Ni(OH)₂ band evolution [9] and operando study of coking in SOFCs [8]

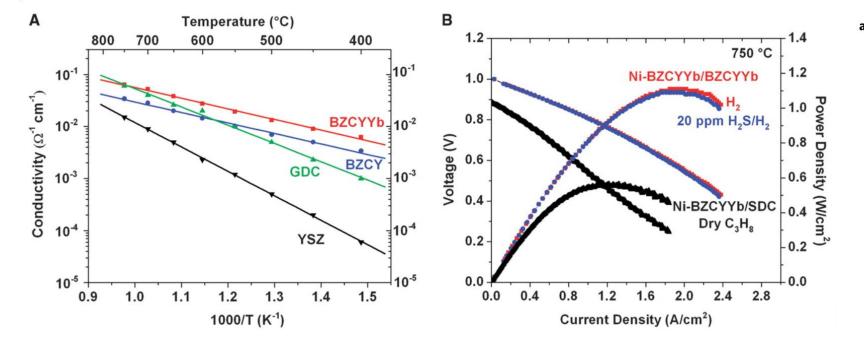


Reversible LiO₂ formation in SnO₂ nanostructures [3]

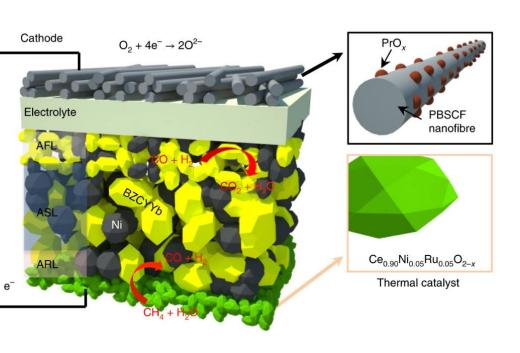
Ni(OH) MOF/graphene composite for supercapacitors [2]

Reversible Solid Oxide Cells

The improvement of fuel cells require **development of catalysts to enable** low-temperature operation and protection against contamination



Development of new electrolyte materials (BZCYYb) [4]

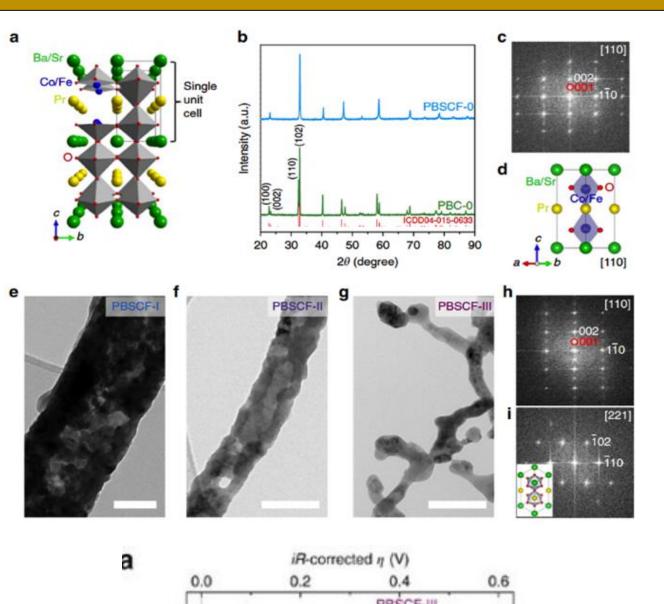


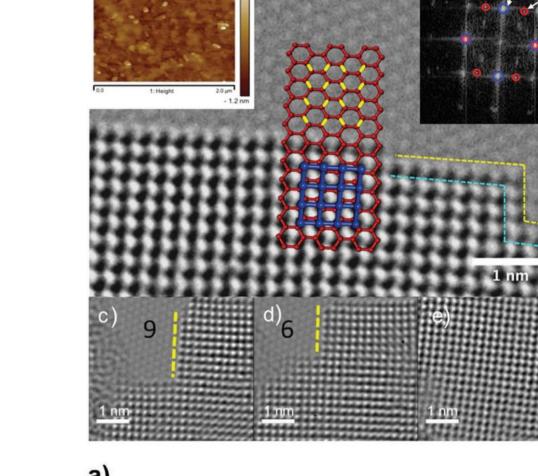
Low temperature methane operation [5]

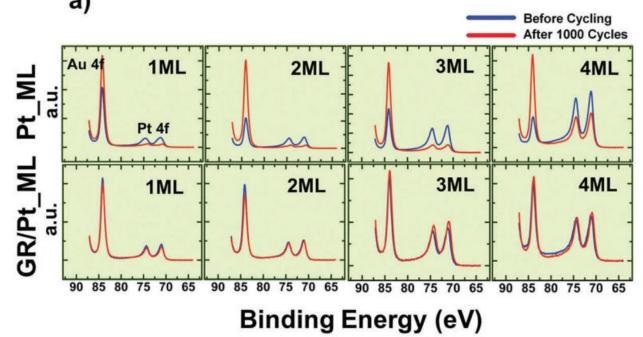
12k

Raman spectroscopy is one of the few techniques that are capable of probing and mapping electrode surfaces under practical operating conditions, allowing direct correlation between the microscopic features and the electrochemical performance of an electrode subject to an applied voltage or current in a wide range of electrochemical environment at temperatures up to 650°C.

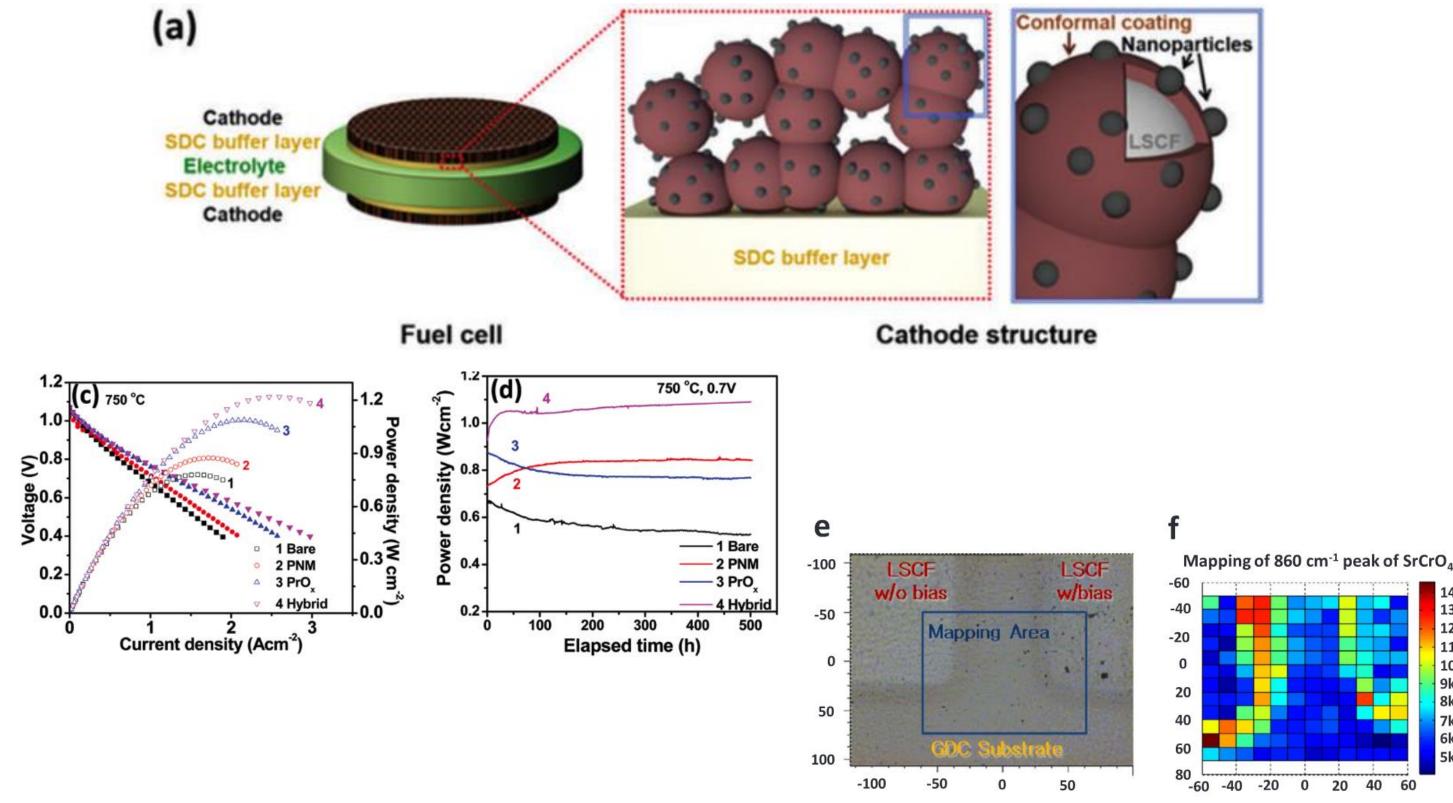
Electrocatalysis and Low Temperature Fuel Cells



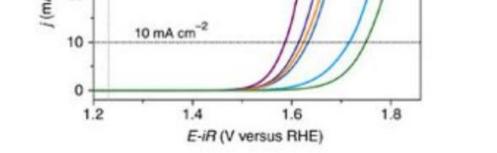








PrNi_{0.5}Mn_{0.5}O₃ coated La_{0.6}Sr_{0.4}Co_{0.2}Fe_{0.8}O₃ (LSCF) for enhanced Oxygen Reduction [6] and analysis and mapping of Cr poising using Raman Spectroscopy [7]



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 $PrBa_{0.5}Sr_{0.5}Co_{1.5}Fe_{0.5}O_{5+\delta}$ nanofibers for OER [11]

Pt/graphene hybrid for ORR [12]

References

- Huang, L.; Chen, D.; Ding, Y.; Feng, S.; Wang, Z. L.; Liu, M. Nano Letters 2013
- Jiao, Y.; Zhao, B.; Liu, M.; Walton, K. S., ACS Applied Energy Materials 2019.
- Hu, R.; Liu, M., Energy & Environmental Science 2016
- Yang, L.; Wang, S.; Blinn, K.; Liu, M.; Liu, Z.; Cheng, Z.; Liu, M., Science 2009 4.
- Chen, Y.; deGlee, B.; Liu, M., *Nature Energy* **2018**
- 6. Chen, Y.; Crumlin, E. J.; Yang, C.; Liu, J.; Yildiz, B.; Liu, M., Energy & Environmental Science 2017
- Chen, Y.; Zhao, B.; Murphy, R.; deGlee, B.; Liu, J.; Liu, Nano Energy 2018
- 8. Li, X.; Blinn, K.; Chen, D.; Liu, M., Electrochemical Energy Reviews 2018
- 9. Chen, D.; El-Sayed, M. A.; Liu, M., *Chemistry of Materials* **2015**
- 10. Lai, S. Y.; Liu, M.; Alamgir, F. M., ChemSusChem 2014
- 11. Zhao, B.; Xiong, X.; Liu, M., Nature Communications 2017
- 12. Abdelhafiz, Alamgir, F. M., *Energy & Environmental Science* **2018**